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f13 f14 Li Na K Be Mg 20 Ca 88 18. Into how many blocks is the periodic table divided? 19.

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How are nonmetals designated in this periodic table? \_\_\_boxes are tinted yellow\_\_\_\_\_ 16. What can be said about the electron configurations of all the elements in a group? \_their valence electron configurations are identical\_\_\_\_\_ The s-, p-, d-, and f-Block Elements . 1. What are the four sections, or blocks, of the periodic table?

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Using the Periodic Table 1. Identify the number of valence

electrons in each of the following elements. a. Ne d. Mg f. Cl h. Si

2. Identify the energy level of the valence electrons in each of the

following elements. a. c. d. e. H Ar 1 3. Use the periodic table to

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write the electron configurations (using noble gas notation) for each of the ...

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What do chocolate chip cookies, chemistry and logic have in

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common? They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table. This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery. The two activities in this unit provide students with information that they have to arrange in organized charts. In the process of creating the arrangements, students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification. This dynamic unit meets national science standards in seven teaching and content areas. Bring the periodic table to life with this hands-on, minds-on unit. Book jacket.

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Prentice Hall Physical Science: Concepts in Action helps students make the important connection between the science they read and what they experience every day. Relevant content, lively explorations, and a wealth of hands-on activities take students' understanding of science beyond the page and into the world around them. Now includes even more technology, tools and activities to support differentiated instruction!

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